

Diffusion and adsorption of N₂ and C₂H₆ in ZIF-8 MD and MC simulations

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Molecular Dynamics (MD) and Gibbs-Ensemble Monte-Carlo (GMC) simulations give insight into the adsorption and diffusion properties of N₂ and C₂H₆ molecules adsorbed in ZIF-8 [1]. The framework flexibility plays an important role in achieving correct transport properties [2]. Particularly, for N₂ a gate opening effect could be observed. The window sizes were determined by the orientation of the imidazolate linkers and a transition from a closed structure at low N₂ loading (approximately 0.5–18.4 N₂/cage), to an open one, at high loading (approximately 20–30 N₂/cage) was found (see Fig. 1). This is in agreement with [3]. The results of MD simulations and of experiments for N₂ and C₂H₆ are compared. The simulation results illustrate that the transition of the orientation of imidazolate linkers of the ZIF-8 framework for both, 4 and 6-membered rings were induced by N₂ guest molecules.

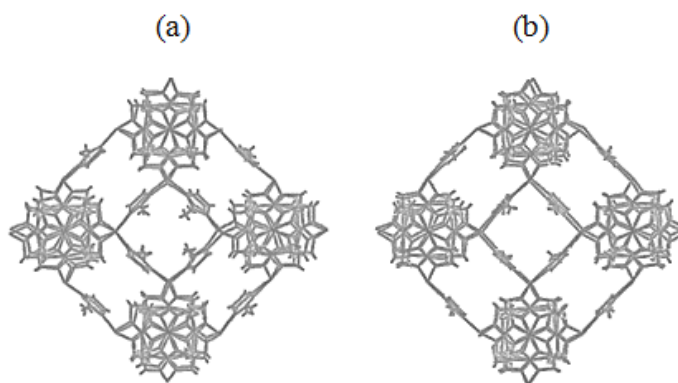


Figure 1: ZIF-8 window structures at (a) low loading and (b) high loading of N₂ guest molecules.

References

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